

TD3 Distribution Canonique

Le système est en contact thermique avec un 'thermostat' qui lui impose de l'extérieur sa température

Ex3) Paramagnétisme de Langevin

N molécules, mt magnétique $\vec{\mu}$, chp extérieu $\vec{B} = B \vec{e}_z$: $w = -\vec{\mu} \cdot \vec{B}$

1) 2 situations $\begin{matrix} \vec{\mu} \uparrow \\ \vec{\mu} \downarrow \end{matrix} \begin{matrix} \uparrow \vec{B} \\ \uparrow \vec{B} \end{matrix} \Rightarrow \begin{matrix} w_{\uparrow} = -\mu B \\ w_{\downarrow} = -(-\mu B) = \mu B \end{matrix}$

$$P_{\uparrow} = \frac{e^{-\beta w_{\uparrow}}}{e^{-\beta w_{\uparrow}} + e^{-\beta w_{\downarrow}}} = \frac{e^{-\beta \mu B}}{e^{-\beta \mu B} + e^{\beta \mu B}}$$

avec $\beta = \frac{1}{k_B T}$

$$P_{\downarrow} = \frac{e^{-\beta \mu B}}{e^{\beta \mu B} + e^{-\beta \mu B}}$$

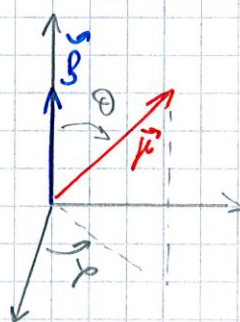
\Rightarrow Valeur moyenne du mt magnétique $\langle \Pi_3(T) \rangle = N(P_{\uparrow} \mu + P_{\downarrow} (-\mu))$

$$\langle \Pi_3(T) \rangle = N \mu \left(\frac{e^{-\beta \mu B} - e^{\beta \mu B}}{e^{-\beta \mu B} + e^{\beta \mu B}} \right)$$

on pose $x = \frac{\mu B}{k_B T} = \beta \mu B$ sans dimension

$$\langle \Pi_3(T) \rangle = N \mu \frac{e^x - e^{-x}}{e^x + e^{-x}} = N \mu \operatorname{th} \left(\frac{\mu B}{k_B T} \right)$$

2) a) μ selon Θ et $\Theta + d\Theta$



$w = -\vec{\mu} \cdot \vec{B} = -\mu B \cos \Theta$
système sphérique

W indep de φ

proba dP que $\vec{\mu}$ pointe dans l'angle solide $d\Omega = \sin \Theta d\Theta d\varphi = 2\pi \sin \Theta d\Theta$
 \equiv proportionnelle à $e^{-\frac{W}{k_B T}}$
 (mbre d'états répondant aux cdt's fixés par $d\Omega$)

$$\Rightarrow dN(\theta) = cste e^{-\frac{W}{k_B T}} 2\pi \sin\theta d\theta$$

$$= cste e^{\frac{\mu_B \cos\theta}{k_B T}} 2\pi \sin\theta d\theta$$

Détermination de $cste$

$$N = cste \int_0^\pi e^{\frac{\mu_B \cos\theta}{k_B T}} 2\pi \sin\theta d\theta$$

$$\begin{aligned} \cos\theta &= u \\ -\sin\theta d\theta &= du \end{aligned}$$

$$= cste \int_{-1}^1 e^{\frac{\mu_B u}{k_B T}} (-2\pi) du = \frac{2\pi cste k_B T}{\mu_B} \left[e^{\frac{\mu_B u}{k_B T}} \right]_{-1}^1 = \frac{2\pi cste k_B T}{\mu_B} \left(\frac{e^{\frac{\mu_B}{k_B T}} - e^{-\frac{\mu_B}{k_B T}}}{2} \right)$$

$$\Rightarrow cste = \frac{\mu_B N}{4\pi k_B T \operatorname{sh}\left(\frac{\mu_B}{k_B T}\right)}$$

Soit $\Rightarrow dN(\theta) = \frac{\mu_B N}{4\pi k_B T \operatorname{sh}\left(\frac{\mu_B}{k_B T}\right)} e^{\frac{\mu_B \cos\theta}{k_B T}} 2\pi \sin\theta d\theta$

$$= \frac{\mu_B N}{2 k_B T} \frac{1}{\operatorname{sh}\left(\frac{\mu_B}{k_B T}\right)} e^{\frac{\mu_B \cos\theta}{k_B T}} \sin\theta d\theta \quad \text{ou} \quad dP(\theta) = \frac{1}{2} \frac{\mu_B}{k_B T} \frac{1}{\operatorname{sh}\left(\frac{\mu_B}{k_B T}\right)} e^{\frac{\mu_B \cos\theta}{k_B T}} \sin\theta d\theta$$

$$b) \langle \Pi_z(T) \rangle = \int_0^\pi N \mu_z dP(\theta) = \int_0^\pi N \left(\frac{\mu_B \cos\theta}{k_B T} \right) \frac{1}{2} \frac{\mu_B}{k_B T} \frac{1}{\operatorname{sh}\left(\frac{\mu_B}{k_B T}\right)} e^{\frac{\mu_B \cos\theta}{k_B T}} \sin\theta d\theta$$

$$= \frac{N \mu_B^2}{2 k_B T} \frac{1}{\operatorname{sh}\left(\frac{\mu_B}{k_B T}\right)} \int_0^\pi e^{\frac{\mu_B \cos\theta}{k_B T}} \underbrace{\cos\theta}_{u} \sin\theta d\theta$$

$$c) \quad u = \cos\theta$$

$$v = \frac{k_B T}{\mu_B} e^{\frac{\mu_B \cos\theta}{k_B T}}$$

$$du = -\sin\theta d\theta$$

$$dv = e^{\frac{\mu_B \cos\theta}{k_B T}} \sin\theta d\theta$$

INTEGRATION PAR PARTIE

$$\langle \Pi_z(T) \rangle = \frac{N \mu_B^2}{2 k_B T} \frac{1}{\operatorname{sh}\left(\frac{\mu_B}{k_B T}\right)} \left[\left[\cos\theta \frac{k_B T}{\mu_B} e^{\frac{\mu_B \cos\theta}{k_B T}} \right]_0^\pi - \frac{k_B T}{\mu_B} \int_0^\pi e^{\frac{\mu_B \cos\theta}{k_B T}} \sin\theta d\theta \right]$$

$$= \frac{N \mu_B^2}{2 k_B T} \frac{1}{\operatorname{sh}\left(\frac{\mu_B}{k_B T}\right)} \left[\frac{k_B T}{\mu_B} \left(e^{\frac{\mu_B}{k_B T}} + e^{-\frac{\mu_B}{k_B T}} \right) - \frac{k_B T}{\mu_B} \left(e^{\frac{\mu_B}{k_B T}} - e^{-\frac{\mu_B}{k_B T}} \right) \right]$$

fonction de Langevin.

$$= N \mu_B \left[\frac{\operatorname{ch}\left(\frac{\mu_B}{k_B T}\right)}{\operatorname{sh}\left(\frac{\mu_B}{k_B T}\right)} - \frac{k_B T}{\mu_B} \right] = N \mu_B \left[\operatorname{coth}\left(\frac{\mu_B}{k_B T}\right) - \frac{k_B T}{\mu_B} \right] = N \mu_B \mathcal{L}\left(\frac{\mu_B}{k_B T}\right)$$

$\mathcal{L}(x) = \operatorname{coth} x - \frac{1}{x}$

3) Comparaison $\langle \Pi_3(T) \rangle$ et $\langle \Pi'_3(T) \rangle$ à 1^{ère} température ③

$$\Leftrightarrow \frac{\mu B}{k_B T} = x \ll 1$$

$$e^x \approx 1 + x + \frac{x^2}{2!} + \frac{x^3}{3!} + \frac{x^4}{4!}$$

$$\coth x = \frac{e^x + e^{-x}}{e^x - e^{-x}} \approx \frac{(1 + x + \frac{x^2}{2} + \frac{x^3}{6} + \frac{x^4}{24}) + (1 - x + \frac{x^2}{2} - \frac{x^3}{6} + \frac{x^4}{24})}{(1 + x + \frac{x^2}{2} + \frac{x^3}{6} + \frac{x^4}{24}) - (1 - x + \frac{x^2}{2} - \frac{x^3}{6} + \frac{x^4}{24})}$$

$$\approx \frac{(2 + x^2)}{2x + \frac{x^3}{3}} = \frac{1 + \frac{x^2}{2}}{x(1 + \frac{x^2}{6})} = \frac{1}{x} \frac{(1 + \frac{x^2}{2})(1 + \frac{x^2}{6})^{-1}}{1} \approx \frac{1}{x} \left(1 + \frac{x^2}{2}\right) \left(1 - \frac{x^2}{6}\right) \approx \frac{1}{x} \left(1 + \frac{x^2}{3}\right) = \frac{1}{x} + \frac{x}{3}$$

soit $\mathcal{L}(x) = \coth x - \frac{1}{x} \approx \frac{x}{3}$

$$\langle \Pi'_3(T) \rangle = \frac{1}{3} N \mu^2 \frac{B}{k_B T}$$

$$\langle \Pi_3(T) \rangle = N \mu \frac{1}{\coth(\frac{\mu B}{k_B T})} \approx N \mu \coth\left(\frac{\mu B}{k_B T}\right) \approx N \mu \left(\frac{1}{x} \left(1 - \frac{x^2}{3}\right)\right)^{-1} = N \mu x = N \mu^2 \frac{B}{k_B T} = 3 \langle \Pi'_3(T) \rangle$$

Ex 1) Gas parfait dans l'ensemble canonique
 équilibre thermodynamique à T, N molécules, masse m, V

1) Fonction de partition d'une molécule: \mathcal{Z} \Rightarrow q^{te} sans dimension

$$\mathcal{Z} = \iiint \iiint e^{-\frac{p^2}{2mk_B T}} \frac{d^3 \vec{r}}{h^3} d^3 \vec{p} = \frac{1}{h^3} \underbrace{\iiint d^3 \vec{r}}_V \iiint e^{-\frac{(p_x^2 + p_y^2 + p_z^2)}{2mk_B T}} dp_x dp_y dp_z$$

$$= \frac{V}{h^3} \left[\int_{-\infty}^{+\infty} e^{-\frac{p_x^2}{2mk_B T}} dp_x \right]^3$$

$$= \frac{V}{h^3} (2\pi m k_B T)^{3/2}$$

$$\int_0^{\infty} e^{-ax} dx = \frac{1}{a}$$

$$a = \frac{1}{2mk_B T}$$

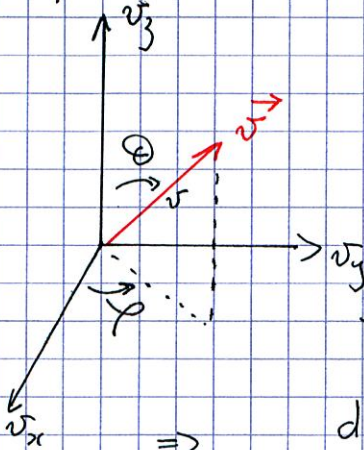
2) Fonction de partition du gaz parfait: Z N molécules indiscernables

$$Z = \frac{\mathcal{Z}^N}{N!} \Rightarrow \text{approximation de PB}$$

Ex 5] Pompage cryogénique

N_0 molécules m ds V_0 contact avec T
 $PV = N_0 k_B T$

1) Ss espace des vitesses



$$dP(\vec{v}) = \underbrace{c}_{\substack{\text{densité de proba} \\ \text{en fct de distribution de } \vec{v}}} e^{-\beta \frac{mv^2}{2}} d^3\vec{v}$$

$$= \frac{dN(\vec{v})}{N_0} \leftarrow \text{nbre de molécules ayant } \vec{v} \text{ à } d^3\vec{v} \text{ près}$$

soit $dN(\vec{v}) = N_0 c e^{-\beta \frac{mv^2}{2}} d^3\vec{v}$

Calcul de $c \Rightarrow \iiint dN(\vec{v}) = N_0 \Rightarrow c \int_0^\infty v^2 e^{-\beta \frac{mv^2}{2}} dv \int_0^\pi \sin\theta d\theta \int_0^{2\pi} d\varphi = 1$

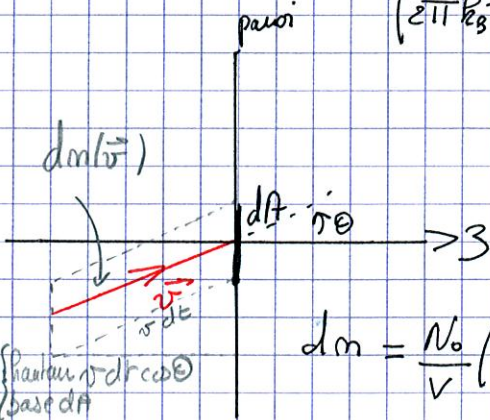
$$4\pi c \int_0^\infty v^2 e^{-\beta \frac{mv^2}{2}} dv = 1$$

$$\frac{1}{4} \sqrt{\frac{8\pi k_B T}{m^3}}$$

Finalemnt $\Rightarrow dN(\vec{v}) = N_0 \left(\frac{m}{2\pi k_B T} \right)^{3/2} v^2 e^{-\frac{mv^2}{2k_B T}} dv \sin\theta d\theta d\varphi$

$$\pi c \sqrt{\frac{8\pi k_B T}{m^3}} = 1 \Rightarrow c = \left(\frac{m}{2\pi k_B T} \right)^{3/2}$$

2/



$$\frac{dm(\vec{v})}{N_0} = \frac{(dA r dt \cos\theta)}{v} dP(\vec{v})$$

$$\Rightarrow dm(\vec{v}) = N_0 \left(\frac{m}{2\pi k_B T} \right)^{3/2} v e^{-\frac{mv^2}{2k_B T}} dv dA dt \sin\theta \cos\theta d\theta d\varphi$$

3/ cylindre hauteur $r dt \cos\theta$ base dA

$$dm = \frac{N_0}{v} \left(\frac{m}{2\pi k_B T} \right)^{3/2} dA dt \int_0^\infty v^3 e^{-\frac{mv^2}{2k_B T}} dv \int_0^{2\pi} d\varphi \int_0^\pi \sin\theta \cos\theta d\theta$$

$$= \frac{N_0}{v} \left(\frac{m}{2\pi k_B T} \right)^{3/2} dA dt \times \frac{4k_B T}{2m^2} \times \frac{2\pi}{2\pi} \times \frac{1}{2} = \frac{N_0}{v} dA dt \left(\frac{k_B T}{2\pi m} \right)^{1/2}$$

refroidissement de $dA \Rightarrow$ condensation $\Rightarrow P \downarrow$: pompage cryogénique

4) $N(t=0) = N_0$ $dN(t)$ nombre de molécules qui frappent dA pendant dt

$$\frac{dm}{N_0} = - \frac{dN(t)}{N(t)} = \frac{1}{V} dA dt \left(\frac{p_0 T}{2\pi m} \right)^{1/2} \Rightarrow dN(t) = - \frac{N(t)}{V} dA dt \left(\frac{p_0 T}{2\pi m} \right)^{1/2}$$

ou encore $\frac{dN(t)}{N(t)} = -d dt \Rightarrow N(t) = N_0 e^{-dt}$

avec $d = \frac{dA}{V} \left(\frac{p_0 T}{2\pi m} \right)^{1/2}$

Pression $P(t)$ idéale $P(t) V = N(t) p_0 T$ (GP)

$$\Leftrightarrow P(t) = \frac{N(t)}{V} p_0 T = \frac{N_0}{V} e^{-dt} p_0 T = P_0 e^{-dt}$$

AN. $V = \frac{4}{3} \pi R^3 \cdot dA = 10^{-4} \text{ m}^2$
 $= \frac{4}{3} \pi 10^{-3} \text{ m}^3$

$$\Rightarrow t = \frac{1}{d} \ln \frac{P_0}{P(t)} = \frac{V}{dA} \left(\frac{2\pi m}{p_0 T} \right)^{1/2} \ln \frac{P_0}{P(t)}$$

$$t = 33 \text{ s}$$